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1. Which is known as carbolic acid?

- A. Phenol**
- B. Ethanol
- C. Acetic acid
- D. Oxalic acid

2. Name the most common indicator used in a laboratory?

- A. Methyl Orange
- B. Red Litmus
- C. Universal Indicator**
- D. Phenolphthalein

3. Name an element which is common to all acids?

- A. Sulphur
- B. Chlorine
- C. Nitrogen
- D. Hydrogen**

4. What type of reaction takes place when an acid dissolves in water:

- A. Exothermic**
- B. Endothermic
- C. Substitution
- D. Double-displacement

5. In pH, the p stands for

- A. potential**
- B. proton
- C. primitve
- D. process

6. On a pH scale, we can measure pH from

- A. 0-7
- B. 7-14
- C. 0-14**
- D. None of the above

7. Name an acid which contains both oxygen and hydrogen?

- A. Oxyacid**
- B. Hydra acid
- C. Dilute acid
- D. Concentrated acid

8. The higher the hydronium ion concentration of a solution

- A. **the lesser is the pH**
- B. the higher is the pH
- C. there's no change in pH
- D. pH first increases then decreases

9. What happens when an acid reacts with metal oxide?

- A. **Salt and water is formed**
- B. Metal hydride is formed
- C. Oxyacid will be formed
- D. Salt and Hydrogen gas is formed

10. Which of the following acid is present in Tomato?

- A. Phosphoric acid
- B. Tartaric acid
- C. **Oxalic acid**
- D. Lactic acid